Europäisches Patentamt

European Patent Office

Office européen des brevets



(11) EP 0 987 023 A1

(12)

EUROPEAN PATENT APPLICATION

(43) Date of publication: 22.03.2000 Bulletin 2000/12

(51) Int. Cl.7: A61K 31/315, A61K 31/19

(21) Application number: 98610026.1

(22) Date of filing: 17.08.1998

(84) Designated Contracting States:
AT BE CH CY DE DK ES FI FR GB GR IE IT LI LU
MC NL PT SE

Designated Extension States:

AL LT LV MK RO SI

(71) Applicant:
PANACEA BIOTEC LIMITED
New Dehli 110001 (IN)

(72) Inventors:

 Amarjit, Singh New Dehli-110001 (IN) Rajesh Jain New Dehli-110001 (IN)

 Anil,KumarSingla c/o Ms.Chagumal Santram Punjab (IN)

(74) Representative:

Christiansen, Ejvind et al c/o Hofman-Bang & Boutard, Lehmann & Ree A/S, Hans Bekkevolds Allé 7

2900 Hellerup (DK)

(54) Transition metal complexes of non steroidal anti-inflammatory drugs

(57) A COX $_2$ selective pharmaceutical composition is disclosed. The composition has a potency ratio less than 1 and comprises a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -

(Drug)₂ Zn.nH₂O.

The drug in the above formula is Naproxen in a suitable pharmaceutical base/carrier or diluent.

EP 0 987 023 A1

Description

INTRODUCTION

[0001] The present invention relates to a novel pharmaceutical composition comprising Naproxen and a transition metal. The Novel drug metal complex is characterised in having increased efficacy and manifolds better tolerability than the parent drug and standard NSAIDS. More particularly the invention relates to pharmaceutical complex of Naproxen and Zinc, obtained by reacting Naproxen and (or) one or more salts of Naproxen with Zinc in one or more salt form(s).

10 BACKGROUND OF THE INVENTION

[0002] Naproxen (US patents 3,904,682 & 4,009, 197) is (+) -2-(6-Methoxy-2-haphthyl) propionic acid [I.T. Harrison et al J. Med. Chem., 13 203 (1970)]. It has analgesic, antiinflammatory and antipyretic properties. It is an inhibitor of Cyclo-oxygenase [(Roszkowski et al Pharmacol. Exp. Ther 179, 114 (1971)], [Tomlinson et al Biochem Biophys REs commun 46 552 (1972)]. Both Naproxen and Naproxen Sodium are used in muscoskeletal and joint disorders such as ankylosing spondylitis, osteoarthritis and rheumotoid arthritis, in mild to moderate pain such as dysmenorrhoea, migraine and some musculoskeletal disorders, and in acute gout [R.N. Brogden et al Drugs, 18, 241-277 (1979)]. In some of the conditions the drug has to be administered for long periods of time. Among the other adverse-effects, Gastro-intestinal adverse effects are among the most frequently reported during short and long term treatment with Naproxen. Hence development of the pharmaceutical forms which reduce the adverse effects, in particular the Gastro-intestinal adverse effects is most desirable.

[0003] There are several references of different types of approaches towards improvement of galenic dosage forms of Naproxen such as programmed release (US Patent Nos. 5480650, 4888178), Diffusion-osmotic controlled drug-release (US patent nos. 5543155, 4571333), Liquid-suspension controlled release (US Patent no. 5527545) Systems for improved bio-availability or increased patient compliance e.g Naproxen micro capsules (W.O. Patent 9505166) and Cyclodextrin inclusion compounds (W.O patent. 9504528) have also been reported. Soluble salts of Naproxen have been prepared with N-(2-hydroxy ethyl) pyrrolodone (U.S. patent no. 5206262).

[0004] Drug ligand-metal complexes of several therapeutic compounds have been reported. In several cases the ligand-metal complexes have better therapeutic and/or pharmaceutical properties than the parent ligand. The prior art regarding this aspect for NSAIDS is descussed below.

[0005] An inclusion complex of a NSAID or a pharmaceutically acceptable Salt thereof and a Cyclodextrin, and a physiologically acceptable alkali agent selected from the group consisting of alkali and alkaline earth metal carbonates, bicarbonates, phosphate and hydroxides, and water-soluble amines have been described in WO patent no. 9614839. The alkali agent has been shown to be capable of forming an alkaline diffusion layer around the composition in the gastro-intestinal tract.

[0006] Chen et al (J. Inorg. Biochem. 1992 Aug; 47 (2); 81-7) have described synthesis, characterisation, and anti-inflammatory activity of Naproxen complexes with rare earth (III). Various analyses have indicated presence of 1:3 (metal: ligand) stochiometry and the carboxylate group of Naproxen has been suggested as a bridging ligand to co-ordinate to RE (III) ions. Two pharmacological models were chosen to examine the antiinflammatory activity of Nd(III) complex, which ascertained enhanced anti-inflammatory activity relative to the ligand.

[0007] It has been known now for 60 years that Zinc is essential for animal (including human) life. Approximately 300 enzymes require Zinc for their activities and it has a role in DNA Synthesis, Cell division and protein synthesis (Peasad A, Nutrition 1995, 11, 93-9). The inventors have looked into the possibility of making Zinc complexes of NSAIDs, particularly of Naproxen which have not been reported so far

[0008] Navarro et al (Prostaglandins Leukot. Essent. Fatty Acids, 1994 Jun; 50 (6); 305-10) studied the effect of pretreatment with Zinc acexarnate on Gastrotoxic activities of different NSAIDs (diclofenac, indomethacin, Ketoprofen, naproxen and piroxicam). Zinc acexamate pre-treatment significantly decreased the overall Severity of lessions induced by NSAIDS. The experiments Corroborate the hypothesis that the preservation of the capability to synthesise endogenous prostaglandins is of critical importance in the maintenance of gastric mucosal integrity. The gastro-protective action observed with Zinc acexamate involves alternative mechanisms other than modification of PGE levels.

[0009] US Patent no. 5466824 describes a process for the preparation of a complex of indomethacin and a divalent metal comprising forming a solution by dissolving indomethacin and a salt of said divalent metal in a tertiary amide or cyclic teritory amide, adding a C_{14} alkanol or C_{3-6} Ketone to the solution to precipitate the complex, and separating the precipitated complex from the solution. This art also provides a method for the treatment of inflammation and/or pain in a mammal requiring such treatment, comprising administering to said mammal an anti-inflammatory or analgesically effective amount of a complex of indomethacin and a divalent metal, the complex having the formula (M_2) (Indomethacin₄) (S_n) wherein M is the divalent metal, S is a molecule of tertiary amide or a cyclic tertiary amide, and n is 2 or 3, or of a pharmaceutical composition comprising said complex together with a pharmaceutically acceptable car-

rier, diluent and/or excipient. This US patent further disclosures a complex of indomethacin and a divalent metal, the complex having the formula [M2] [indomethacin4] [Sn], wherein M, S, and n are defined above, and a pharmaceutical composition comprising this complex together with a pharmaceutically acceptable carrier, diluent and/or excipient.

[0010] The zinc-aspirin complex has been reported by A.K. Singla et. al. (International Journal of Pharmaceutics, 108 (1994), 173-185). The complex was reported to have better efficacy and tolerability profile. Further Zinc-indomethacin complex has been reported (A.K. Singla et. al. International Journal of Pharmaceutics, 120 (1995) 145-155. The zincindomethacin complex was more potent than the parent drug and also had a better tolerability profile with respect to ulcerogenic properties.

[0011] U.S. Patent No. 5466824 describes a complex formation of indomethacin with a divalent metal and a tertiary amide or a cyclic tertiary amide while that to C_{14} alkanol or C_{3-6} Ketone. Inclusion of alien molecules in a complex changes the total course of drug development process and substantial data in order to prove safety and efficacy of a new complex is essential. The process of preparation itself in U.S. Patent No. 5466824 is very cumbersome, time comsuming, costly and the presence of new entities besides the drug and the divalent metal may take considerable time and pre-clinical and clinical tests before the drug can be put to usefulness.

[0012] One of the inventors of the group of inventors of the present application has reported the formation of zinc-NSAID complexes. The zinc-indomethacin has been shown to be better than the parent molecule. Similarly, the aspirinzinc complex has been shown to be better than the parent molecule. However both these NSAIDs drugs have their own disadvantages in long term use. Indomethacin has low therapeutic index while aspirin has to be used in very high dosage to produce anaphylactic reactions. Although the zinc salts complexes are better, they may not be useful in therapeutics.

[0013] Besides indomethacin is not commonly used for therapy as an analgesic or antipyretic because of high incidence and severity of side effects associated with long term administration. It is reserved for special use in acute gout and severe ankylosing spondylitis and osteoarthritis. This is evident from ED50, LD50, and therapeutic index, values of NSAIDs as collected herein below.

25

20

30

35

40

S.No.	Drug	ED ₅₀ (mg/kg)	LD ₅₀ (mg/kg)	Therapeutic Index
1.	Nimesulide	1.25	324	260
2.	Naproxen	2.10	395	190
3.	lbuprofen	13.5	923	68
4.	Diflumidone	38.0	750	20
5.	Flufenamic acid	14.7	249	17
6.	Phenylbutazone	20.5	406	14
7.	Acetylsalicylic acid	135.0	1520	14
8.	Indomethacin	2.95	210	7

[0014] Early approaches for improvement in efficacy and side effect profile were targetted with respect to drugs with low therapeutic index and poor safety margin.

[0015] NSAIDs can be ranked by concentration (μ g/ml) of drug necessary to achieve 50% inhibition (IC₅₀) of COX₂ divided by IC50 for COX1, and the values below 1 indicate selectivity for COX2. The similarities in rates of g.i. complications were found to be as striking as the difference in ratio of COX1 and COX2 inihitory action as illustrated below.

[0016] The data as per the reference "Drugs of Today Vol. 32 Suppl. D1996 Page 5" as illustrated herein below shows that only four compound out of which Naproxen and Diclofenac are examples are selective for COX2 potency ratio < 1.

Among the many available NSAIDs, indomethacin and aspirin are COX₁ selective with potency ratio upto 100.

Drug	Cyclooxygenase 2/Cyclooxygenase 1 ratio
Piroxicam	250

(continued)

Drug	Cyclooxygenase 2/Cyclooxygenase 1 ratio
Tolmetin	175
Acetylsalicylic acid	166
Sulindac	100
Indomethacin	60
Tolfenamic acid	16.7
Ibuprofen	15
Paracetamol	7.4
Sodium salicylate	2.8
Flurbiprofen	1.3
Carprofen	1
Meloxicam	0.8
Diclofenac	0.7
Naproxen	0.6
Nimesulide	0.1

[0017] On the other hand while Naproxen has a comparative COX_2/COX_1 ratio (0.6) which is much lesser in comparision to other NSAIDs such as aspirin (166) and indomethacin (60).

[0018] After careful and planned experimentation and with a quest to a improve upon the drug candidates with relatively high therapeutic index and relatively high safety margin, the inventors of the present invention conceived, planned and executed a research project to evaluate the zinc complexation on safety and efficacy of such drug canditates by taking examples of Naproxen and diclofenac.

[0019] The compositions of the present invention are therapeutically useful efficacious and safe, simple to manufacture and the process of manufacture is not time consuming. Besides the entire process and the drug manufactured thereby is not expensive.

SUMMARY OF THE INVENTION

5

10

15

20

25

[0020] The present invention discloses a novel pharmaceutical COX₂ selective composition having potency ratio less than 1 of Naproxen and/or one or more salts and/or adducts of Naproxen and zinc in one or more salt forms.

DETAILED DESCRIPTION OF THE INVENTION

[0021] The inventors have synthesised Zinc-diclofenac complex. This complex was found to have a more potent antiinflammatory activity than diclofenac but it showed more toxicity in experimental animals.

[0022] In accordance with the present invention the pharmaceutical composition comprising Naproxen and zinc can be represented by the general formula (Drug)₂Zn.nH₂O. In other words two molecules of the drug are reacted with one molecule of zinc in the presence of water.

[0023] The salts and adducts of Naproxen and Diclofenac are selcted from the group optically pure Naproxen, Naproxen-Lysine salt, Naproxen-Cyclodextrin Adduct, Diclofenac Potassium, Diclofenac Diethylammonium, Diclofenac Epolamine and Diclofenac Tromethamine.

[0024] The salts of zinc are selected from zinc carbonate, zinc acetate, zinc sulphate, zinc chloride and zinc glucomate.

[0025] In accordance with another aspect of the present invention, is disclosed the synthesis of Naproxen-Zinc complex. After careful and planned experimentation on studies of pharmacological properties of the metal-ligands prepared it has been surprisingly found that Naproxen-Zinc complex have Synergistically higher anti-inflammatory and analgesic activities when compared to the parent drugs, Zinc alone and physical mixture of the drug and Zinc. There is a trend that as anti-inflammatory and analgesic properties of a NSAID compound increase, the side effects particularly the gas-

tric-irritation also increases. The phenomena is attributed to inhibition of endogenous PEG_2 in gastric mucosa. Moreover it has been further surprisingly found that Naproxen-Zinc complex has least gastrotoxic side effects measured by ulcer index although it has a very high anti-inflammatory and analgesic activity.

[0026] In another embodiment of the present invention the composition is disclosed where the Naproxen is converted into a therapeutically safe and efficacious form i.e. Naproxen-zinc just before administration through oral or other construable routes. Such compositions are very simple to prepare, cost effective and useful. Such compositions have two parts. In one part Naproxen is present in a suitable pharmaceutical carrier and the other part contains a titrated reactive form of zinc with suitable pharmaceutical carrier. In such a composition both the parts are separately reacted in a suitable amount of medium or are present in a admixture form which is reacted just prior to administration.

[0027] The Naproxen-zinc complex was prepared by reacting Zinc Carbonate and Naproxen Sodium in primarily aqueous solvent, which may contain one or more polar organic solvent, in a molar ratio such that two moles of Naproxen react with one mole of Zinc. The complex thus formed was separated and dried.

[0028] The dried compound was subjected to elemental analysis, Infra-Red Spectroscopy Nuclear Magnetic Resosance spectroscopy and Differential Scanning Calorimetory (DSC). The formation of Naproxen Zinc complex was proved by comparing IR scans and DSC graphs of the complex with individual compound separately Examination of the IR-spectra of Naproxen and the synthesised Zinc-Naproxen complex revealed a definite shift in absorption for the carboxyl group and the disappearance of carboxyl O-H stretching and bending. The shift occurred in the direction of longer wavelength, indi cating that the carboxyl group of Naproxen is strongly involved in complexation with Zinc. Comparison of the H-NMR spectra of the two compounds did not show much difference as in the case of Naproxen due to exchange broadening involving the water in the solvent. The formation of the Zinc complex was further confirmed by C¹³-NMR spectra which revealed a strong shift in the absorption of -COOH and slight shifts in the case of other carbon atoms. DSC curves for Naproxen, the physical mixture of Naproxen and Zinc sulphate, and Zinc-Naproxen complex, showed that the endothermic peak of Naproxen disappeard completely in the Zinc complex with the appearance of two new endothermic peaks. The first peak is due to loss of water of crystallisation and the second sharper peak is due to melting with decomposition of the Zinc complex.

[0029] Diclofenac, Naproxen, Diclofenac-Zinc and Naproxen Zinc along with normal saline (as control) and Indomethacin (as standard Anti-inflammatory drug) were subjected to the pharmacological screening tests.

[0030] The anti-inflammatory of the compounds was tested by Carageenan - induced rat paw edema and ulcerogenic effect was evaluated by finding ulcer index.

[0031] The findings of the study are summarised in table 1 & 2 and depicted in fig. 1 & 2.

[0032] The invention provides Naproxen-Zinc complex and a process for manufacture thereof not reported earlier and the initial pharmacological testing indicates a strong possibilities that this complex may be developed as a therapeutically useful compound with reduced dry dose and better tolerability w.r.t gastric irritation.

95 PREPARATION OF ZINC-NAPROXEN COMPLEX

[0033] Naproxen (69.1 g, 0.3 ml) was dissolved in a solution of sodium bicarbonate (25.2 g, 0.3 ml) in water (500 ml) and filtered (pH 8.14). To this solubility was addeed slowly and with constant stirring a solution of zinc acetate dihydrate (35 g, 0.15 ml) in water (180 ml). Immediate precipitation occurred and the precipitates were filtered, washed with a minimum quantity of absolute alcohol and then with cold water, and dried under vaccum to a constant weight to give zinc-naproxen complex (yield, 69.75g, 83.1%); m.p. 222 - 224°C.

CHARACTERISATION OF ZINC-NAPROXEN COMPLEX

[0034] Comparison of the IR spectra of naproxen and its zinc complex showed the disappearance of Carboxyl OH stretching (3166cm⁻¹) in the IR spectrum of the complex occurrence of strong shift in the absorption due to C = O group stretching (1727 to 1689 cm⁻¹) and the shift in -CH-COOH stretch from 1175 to 160 cm⁻¹ also indicating the possibility of host guest interchanging at this site. The displacement occurred in the directions of longer wavelength indicating that the carboxyl group of naproxen is strongly involved in complexation with zinc. Donation of electron to metal produces lower excitation states and therefore shifts to longer wavelengths (William et al., 1976)

[0035] The inventors effected the experiments to test the anti-inflammatory activity and toxicity by evaluating the ulcer index employing Naproxen-Zinc complex and Diclofenac-Zinc complex synthesised by them. It has been suprisingly found by the inventors that the anti-inflammatory activity of the Naproxen-Zinc and Diclofenac-Zinc complex were much more than the parent drug. However, Diclofenac Zinc complex suffered from the char acteristic of being toxic, the toxicity being proportional to the increase in its anti-inflam matory activity.

[0036] The invention will now be described with reference to the following examples which are by way of illustration:

Naproxen Zinc equivalent to Naproxen 125 mg

Naproxen Zinc equivalent to Naproxen 250 mg

Lactose

Starch

Purified Talc

Lactose

HPMC

HPMC

Purified Talc

Micro Crystalline Cellulose

Micro Crystalline Cellulose

Micro Crystalline Cellulose

Magnesium Stearate

Magnesium Stearate

PVP

Example 1: Tablet dosage form

Example 2: Tablet dosage form

[0037]

5

[0038]

25

30

35

[0039]

40

[0040]

Example 4: Tablet dosage form

Naproxen Zinc equivalent to Naproxen 500 mg

Naproxen Zinc equivalent to Naproxen 375 mg

609.00 mg

152.25 mg

229.15 mg

30.00 mg

10.00 mg 15.00 mg

0.90 mg

12.00 mg 300.00 mg

304.50 mg

110.50 mg

30.00 mg

45.00 mg

10.00 mg 500.00 mg

456.75 mg

148.25 mg

30.00 mg

15.00 mg 650.00 mg

6

10

15

20

Example 3: Tablet dosage form

45

50

(continued)

Lactose (Directly compressible)	141.00 mg
	750.00 mg

Example 5: Tablet dosage form

[0041]

10

15

Diclofenac Zinc equivalent to Diclofenac 50 mg	56.85 mg
Lactose	90.00 mg
Micro Crystalline Cellulose	30.00 mg
Polyvinylpyrrolidone	7.15 mg
Sodium Starch Glycollate	16.60 mg
	200.00 mg

20

Example 6 : Capsule dosage form

[0042]

25

	Naproxen Zinc equivalent to Naproxen 250 mg	304.50 mg
30	Micro Crystalline Cellulose	133.00 mg
	Magnesium Stearate	7.00 mg
	Sodium Lauryl Sulphate	1.5 mg
		500.00 mg

35

40

[0043] Filled in size '0' hard gelatin capsules shells.

Example 7 : Suspension dosage form

[0044]

45

Naproxen Zinc equivalent to Naproxen 125 mg	152.25 mg
Cane Sugar	1000.00 mg
Glycerine	200.00 mg
Sodium Saccharin	4.00 mg
Dr. Sodium EDTA	8.00 mg
Xanthan GUM	100.00 mg
Pineapple Flavour	4.00 mg
Purified Water q.s. to	5.00 ml

50

Example 8 : Enteric Coating Tablet

[0045]

5

10

15

20

25

30

35

Diclofenac Zinc equivalent to Diclofenac 25 mg	28.42 mg
Lactose	100.00 mg
Micro Crystalline Cellulose	71.58 mg
HPMC phthalate	36.00 mg
Isopropyl Alcohol*	
	236.00 mg

^{*} Lost in process

Example 9 : Chewable Tablet Dosage form

[0046]

Naproxen Zinc equivalent to Naproxen 375 mg	456.75 mg
Sugar Granular	365.25 mg
Mannitol	100.00 mg
Starch	60.00 mg
Mixed Fruit Flavour	10.00 mg
Aspartame	2.00 mg
Magnesium Stearate	6.00 mg
	1000.00 mg

Example 10 : Effervscent Tablet

40 [0047]

45

50

Naproxen	250.00 mg
Zinc Carbonate	80.00 mg
Sodium Chloride	10.00 mg
PEG 400	10.00 mg
	350.00 mg

Example 11 : Effervscent Tablet

[0048]

5

10

Naproxen 500.00 mg Citric Acid 60.00 mg Zinc Carbonate 250.00 mg Sodium Chloride 20.00 mg **PEG 400** 20.00 mg

850.00 mg

15

Example 12 Capsules dosage form

[0049]

20

25

Naproxen-Zinc Cyclodextrin Complex 375.00 mg 115.00 mg Lysine 5.00 mg Magnesium Stearate Sodium Lauryl Sulphate 5.00 mg 500.00 mg

30

35

[0050] Fill in size '0' hard gelatin capsule shells

Example 13. Kit

[0051]

40

45

50

Granules A		
Naproxen	250.00 mg.	
Citric Acid	30.00 mg.	
PEG 400	20.00 mg.	
Zinc Acetate	300.0 mg. 130.00 mg.	
Granules B.	•	
Zinc Acetate	130.00 mg.	
Sodium Bicarbonate	92.00 mg.	
Sodium Chloride	28.00 mg.	

[0052] Dissolve B in a glass of water and add A. Administer.

Example 14. Topical formulation

[0053]

Naproxen Zinc	1% w/v
Preservative	0.005 to 0.5% w/v
Buffer/vehicle	98.5 to 98.995% w/v

[0054] The examples of the formulation should not be construed to limit the scope of the invention. In fact following these examples, any desired pharmaceutical formulation containing Naproxen and zinc can be prepared. The composition of the invention can be in any form commonly employed for administration i.e drink solution, a concentrated drink solution to be diluted before use, solution encapsulated in soft gelatin capsules, solution adsorbed on suitable adsorbents leading to formulations such as tablets, capsules, and granules, the solution freeze dried for oral, topical solution, or Injectable dosage forms or the like.

[0055] Another embodiment of this invention is a kit which comprises one or more pharmaceutically acceptable doses of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and one or more acceptable doses of Zinc in one or more salt forms or a mixture thereof.

TABLE -I
Effects of administration of Indomethacin, Diclofenac Sodium, Naproxen, Naproxen-Zinc and Diclofenac-Zinc on Carageenaninduced paw edema in rats.

Group	а	Dose (mg/Kg i p.)	٥,	INCREASE IN PAW VOLUME (%) 15' 60	W VOLUME (%) 30' 60'	120'	180,	240
Saline	6		27.54±13.72	34.98±6.03	28.9±8.85 57.94±8.7 66.96±8.4	66.96±8.4	137.08±34.24	133.34±9.50
Indomethacin	m	10	11.1±1.59	31.26±11.43	51.8±12.32 52.89±2.13 50.72±1.66	3 50.72±1.66	65.42±15.26	94.25±13.2
Diclofenac Sodium	60	'n	13.9±7.15	30.75±6.46	36.71±5.48 41.17±8.65 54.29±3.89	5 54.29±3.89	109.6±23.27	101.53±8.89
Naproxen	60	4	14.77±6.36	23.64±10.05	39.12±4.02 38.93±9.05 69.67±10.01	5 69.67±10.01	84.22±13.6	99.05±14.7
Naproxen-Zinc	m	20	17.49±5	35.78±12.04	41.66±10.7 36.52±4.99 50.22±16.49	9 50.22±16.49	48.6*±9.88	48.26*±2.91
Diclofenac Zinc	m	12	20.14±10.45	23.46±8.33	15.39±7.4 19.92±8.3* 25.97±11.07* 34.31*±16.44	25.97±11.07*	34.31*±16.44	34.71*±14.13

P < 0.05

TABLE 2

	lex obtained in each group of rats after administra- cin, Diclofenac Sodium, Naproxen, Naproxen Zinc and Diclofenac-Zinc					
Group	n	Dose (mg/kg. p.o)	Ulcer Index			
Saline	3	-	4.5±0.86			
Indomethacin	3	10	0 9± 2.5			
Diclofenac Sodium	3	5	4.6 6± 0.88			
Naproxen	3	4	4.66±1.01			
Naproxen-Zinc	3	20	4.16±0.44			
Diclofenac-Zinc	3	12	8.16±0.83*			

^{*} p < 0.05

20 Claims

5

10

15

25

35

45

50

55

 A COX 2 selective pharmaceutical composition having potency ratio less than 1 comprising a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -

(Drug)₂ Zn.nH₂O

wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

- 2. A composition as claimed in claim 1 wherein the salts and adducts of Naproxen are selected from the group comprising optically pure Naproxen, Naproxen-Lysine salt and Naproxen Cyclosporin adduct.
 - 3. A composition as claimed in claim 1 wherein the salt forms of Zinc are selected from the group comprising Zinc carbonate, Zinc acetate, Zinc sulphate, Zinc chloride and Zinc glucomate.
 - 4. A composition as claimed in claim 1 which is in the form of a tablet, capsule, suspension, enteric coating tablet, chewable tablet, effervescent tablet and a topical formulation.
- 5. A kit comprising one or more pharmaceutically acceptable doses of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and one or more acceptable doses of Zinc in one or more salt forms or a mixture thereof.
 - 6. A method of treating an NSAID-indicated condition or symptom with reduced gastrointestinal side effects, said methods comprising: administering an effective amount of a complex of Naproxen, its salt or adduct, or mixtures thereof, and zinc or a salt thereof, said complex represented by the formula-

(Drug)₂ Zn.nH₂O

wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent.

7. A process for the manufacture of a COX₂ selective pharmaceutical composition having potency ratio less than 1 comprising a complex of Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salt forms or mixtures thereof and represented by the formula -

(Drug)₂ Zn.nH₂O

wherein the drug is Naproxen in a suitable pharmaceutical base/carrier or diluent which comprises mixing Naproxen and/ or one or more salts and/or adducts of Naproxen or mixtures thereof and Zinc in one or more salts

forms or mixtures thereof in the presence of water under conventional conditions of temprature and pressure.

- 8. A process as claimed in claim 7 wherein the salts and adducts of Naproxen are selected from the group comprising optically pure Naproxen, Naproxen-Lysine salt and Naproxen Cyclosporin adduct.
- 9. A process as claimed in claim 7 wherein the salt forms of Zinc are selected from the group comprising Zinc carbonate, Zinc acetate, Zinc sulphate, Zinc chloride and Zinc glucomate.
- 10. A process as claimed in claim 7 wherein Naproxen is dissolved in a solution of Sodium bicarbonate in water and filtered prior to mixing with a mixture of Zinc and salts thereof. 10
 - 11. A process as claimed in claim 7 wherein Zinc in one or more salt forms or mixtures thereof are mixed with Naproxen and its salts or adducts or mixtures thereof under constant stirring.

15

20

25

30

35

40

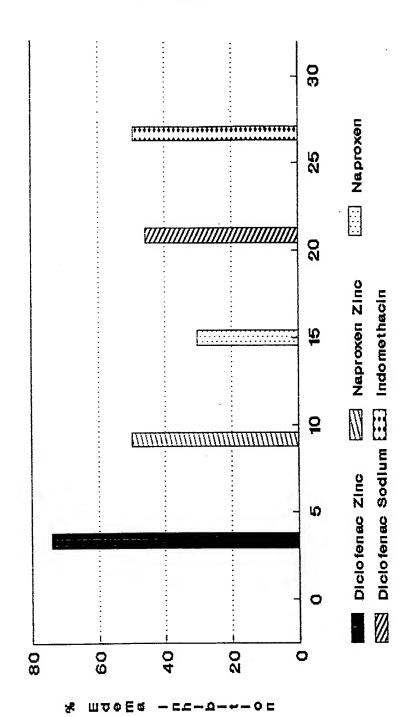
45

50

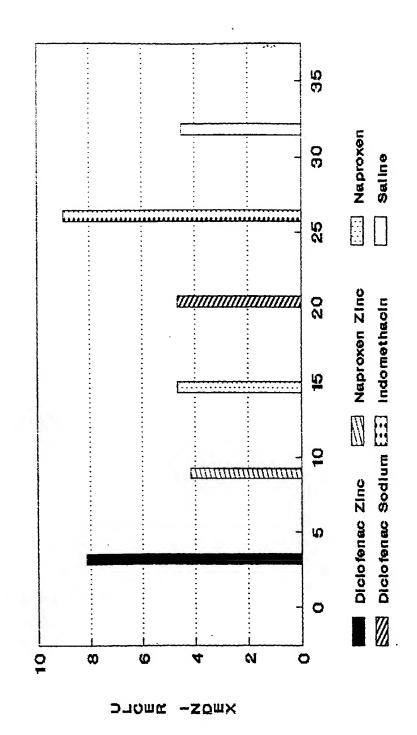
55

Effects of administration of Indomethacin, Diclofenac Sodium, Naproxen-Zinc and FIG-1

Diclofenac-Zinc on Carageenan-induced paw edema in rats.



Values of ulcer index obtained in each group of rats after administration of Indonethacin, Diclofenac Sodium Maproxen, Naproxen Zinc and Diclofenac-Zinc





PARTIAL EUROPEAN SEARCH REPORT

Application Number

which under Rule 45 of the European Patent ConventionEP 98 61 0026 shall be considered, for the purposes of subsequent proceedings, as the European search report

	DOCUMENTS CONSID	ERED TO BE RELEVANT		
Category		ndication, where appropriate.	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
X	EP 0 405 602 A (VIN * abstract * page 3, line 45 ~ example 7 * * page 8, line 41 - claims *	page 4, line 22;	1-11	A61K31/315 A61K31/19
X		"Preparation of y-2-naphthyl)propionate and ulcer inhibitor"	1-11	
D,X	US 4 009 197 A (FRI 22 February 1977 * abstract * * column 3, line 20	ED JOHN H ET AL) - line 48; example 20/	1-11	TECHNICAL FIELDS SEARCHED (Int.CI.6)
The Sear not comp be carried Claims sear Claims sear Reason to Alti	by with the EPC to such an extent that of out, or can only be carried out partial parched completely: parched incompletely: or the limitation of the search: hough claim 6 is dir the human/animal bod	ected to a method of tre ly (Article 52(4) EPC), i l out and based on the a	eatment the	
	Place of search	Date of completion of the search		Examiner
	THE HAGUE	18 January 1999	Hof	f, P
X : part Y : part doc A : tecl O : nor	ATEGORY OF CITED DOCUMENTS iticularly relevant if taken alone iticularly relevant if combined with anor ument of the same category involgical background invertiten disclosure irmediate document	T : theory or principl E : earlier patent do after the filing dat ther D : document cited i L : document cited i	e underlying the current, but public te n the application of other reasons	invention lished on, or

EPO FORM 1503 03.82 (P04C07)



European Patent PARTIAL EUROPEAN SEARCH REPORT Application Number

EP 98 61 0026

	DOCUMENTS CONSIDERED TO BE RELEVANT		CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	•
D,X	US 3 904 682 A (FRIED JOHN H ET AL) 9 September 1975 * abstract * * column 3, line 24 - line 53; example 20 *	1-11	
X	A. RODRIGUEZ DE LA SERNA: "Multicenter Clinical Trial of Zinc Acexamate in the Prevention of Nonsteroidal Antiinflammatory Drug Induced Gastroenteropathy" THE JOURNAL OF RHEUMATOLOGY, vol. 21, no. 5, 1994, pages 927-933, XP002090213 * abstract * * page 928, left-hand column; table 3 *	5	TECHNICAL FIELDS SEARCHED (Int.Cl.6)
A	EP 0 400 558 A (VINAS LAB) 5 December 1990 * the whole document *	1-11	
D,A	SINGLA, ANIL K. ET AL: "Zinc-indomethacin complex: synthesis, physicochemical and biological evaluation in the rat" INT. J. PHARM. (1995), 120(2), 145-55 CODEN: IJPHDE;ISSN: 0378-5173, XP002090214 * the whole document *	1-11	
D,A	SINGLA, ANIL K. ET AL: "Zinc-aspirin complex: synthesis, physicochemical and biological evaluation" INT. J. PHARM. (1994), 108(3), 173-85 CODEN: IJPHDE;ISSN: 0378-5173, XP002090215 * the whole document *	1-11	·
		J	

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 98 61 0026

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

18-01-1999

	Patent document ed in search repo		Publication date		Patent family member(s)		Publication date
EP	0405602	Α	02-01-1991	JP	3128343	A	31-05-19
US	4009197	Α	22-02-1977	US	3904682	 А	09-09-19
				US	4048330	Α	13-09-19
				BE	751445	Α	16-11-19
				CA	960689	Α	07-01-19
				CA	991655	Α	22-06-19
				CH	517690	A	15-01-19
				CH	520644	Α	31-03-19
				CH	520645	A	31-03-19
				CH	537369	Α	13-07-19
				DE	1793825	Α	05-02-19
				DE	1793828	Α	22-04-19
				DE	1668654	Α	15-04-19
				ES	349061	Α	16-08-19
				FR	8487	M	27-07-19
				FR	8494	M	27-07-19
				FR	1587861		03-04-19
				GB	1211134		04-11-19
				HK	26776		21-05-19
				JP	48040726	_	03-12-19
				NL	6800251	A,B	15-07-19
				NL		A	02-03-19
				NL	7512107	A	30-01-19
				SE		В	18-12-19
				US	3896157		22-07-19
				US 	4207241 	A 	10-06-19
US	3904682	Α	09-09-1975	US	4009197	Α	22-02-19
				US	4048330	Α	13-09-19
				BE	751445	Α	16-11-19
				CA		Α	07-01-1
				CA	991655		22-06-19
				CH		Ą	15-01-19
				CH		A	31-03-1
				CH		A	31-03-1
				CH		A	13-07-19
				DE	1793825		05-02-19
				DE		A	22-04-19
				DE		A	15-04-19
				ES	349061		16-08-19
				FR	8487		27-07-19
				FR	8494		27-07-19
				FR	1587861		03-04-19
				GB	1211134		04-11-19
				HK	26776	A	21-05-19

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 98 61 0026

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

18-01-1999

CIG	atent document d in search repo		Publication date		Patent family member(s)	Publication date
US	3904682	Α		JP	48040726 B	03-12-197
		•		NL	6800251 A,B	15-07-196
				NL	7004194 A	02-03-197
				NL	7512107 A	30-01-197
				SE	352069 B	18-12-197
				US	3896157 A	22-07-197
				US	4207241 A	10-06-198
				CH	536805 A	29-06-197
				DE	2013641 A	08-10-197
				FR	2035847 A	24-12-197
				GB	1299295 A	13-12-197
				NL	7004196 A	28-09-197
				ZA	7001191 A	29-09-197
				BE	747812 A	31-08-197
				CA	960668 A	07-01-197
				CH	559161 A	28-02-197
				CH	559162 A	28-02-197
				CH	536803 A	29-06-197
				DE	2005454 A	15-10-197
				FR	2035827 A	24-12-197
				GB	1291386 A	04-10-197
				HK	27176 A	21-05-197
				NL	7004193 A	28-09-197
EP	0400558	 А	05-12-1990	AT	111074 T	15-09-199
				CA	2017746 A	29-11-199
				DE	69012213 D	13-10-199
				DE	69012213 T	22-12-199
				DK	400558 T	13-02-199
				ĬĹ	94548 A	26-05-199
				ĴΡ	1980747 C	17-10-199
				JP	3101647 A	26-04-199
				JP	7002696 B	18-01-199
				PT	94176 A,B	08-02-199
				US	5091547 A	25-02-199

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82